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Section 15-5 Spontaneous Processes <ul><li>A spontaneous process is a physical or chemical change that once begun, occurs with no outside intervention. </li></ul><ul><li>Many spontaneous processes require some energy from the surroundings to start the process. </li></ul> 53.

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### [Solved] Chapter 15, Problem 15.21 - Chemistry: The ...

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Figure 15.4.1 Energy Changes Accompanying the Thermite Reaction Because enthalpy is a state function, the overall enthalpy change for the reaction of 2 mol of Al(s) with 1 mol of Fe<sub>2</sub>O<sub>3</sub>(s) is -851.1 kJ, whether the reaction occurs in a single step ( $\Delta H$  4, shown on the left) or in three hypothetical steps (shown on the right) that involve the successive formation of solid Al<sub>2</sub>O<sub>3</sub> and ...

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